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## ASYMMETRIC MICHAEL REACTION VIA CHIRAL $\alpha$ , $\beta$ -UNSATURATED ALDIMINES

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Effective asymmetric C-C bond formation has only recently been developed in spite of the fact that C-C bond forming reactions play the central role in synthetic organic chemistry. We have reported that 1,4-addition of Grignard reagents to chiral  $\alpha,\beta$ -unsaturated aldimines leads to optically active  $\beta$ -substituted aldehydes in 91-98% enantiomeric excess.<sup>1)</sup> As an extension of this method to the synthesis of a variety of chiral and functionalized organic compounds, we report here the asymmetric Michael addition<sup>2)</sup> of diethyl malonate to the chiral  $\alpha,\beta$ -unsaturated aldimines 1.

It is well known that the Michael addition of malonate system possessing two  $\alpha$ -hydrogen atoms to  $\alpha,\beta$ -unsaturated aldehydes proceeds to give the corresponding adducts in relatively low yields due to the preferential formation of the Knoevenagel condensation products.<sup>3)</sup> After several unsuccessful attempts, however, it has now been found that the Michael adducts of diethyl malonate with  $\alpha,\beta$ -unsaturated aldimines 1 are obtained preferentially by the use of potassium tert-butoxide as a base in THF-EtOH (4.5:1).



In a standard experiment (condition B), a solution of 1 (4.0 mmole) in THF-EtOH (4.5:1)(15 ml) was added to a thick suspension of diethyl potassiomalonate, prepared by adding potassium tert-butoxide (6.0 mmole) to a solution of diethyl malonate (12 mmole) in the same solvent (40 ml), and the whole was stirred at room temperature under  $N_2$  for 45 hr. A usual workup under acidic condition gave the crude product, which was purified by column chromatography on silica gel and the vacuum distillation to pure aldehyde 3. The amino acid ester 4 used as a chiral source was recovered in good yield from the acidic aqueous phase. The results thus obtained are summarized in Table I.

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The optical purity and absolute configuration of 3 were determined by chemical correlation of it with the known 3-methylpentanoic acid 5.

 $\begin{array}{c} H_{3}C \\ H_{1}C \\ H_{1}C \\ H_{1}C \\ CH(COOEt)_{2} \end{array} \xrightarrow{H_{3}C} H_{3}C \\ H_{1}C \\ H_{2}C \\ CH_{2}COOH \\$ 

Since optically pure (5)-(+)-5 is  $(10^{10})_{(benzene)}$  (benzene)  $(10^{10})_{(neat)}$  (neat reported to have  $[\alpha]_{D}^{25}+8.83^{\circ}$  (neat),  $(10^{10})_{(neat)}$  it was determined that optically pure (5)-(+)-3 should have  $[\alpha]_{D}^{20}+16.8^{\circ}$  (benzene).

to $\alpha,\beta$ -unsaturated aldimines 1 3					
run	α,β-unsaturated aldimine	reaction <sup>a)</sup> condition	Chem. Y. <sup>b)</sup> (%)	Opt. Y. <sup>C)</sup> (%)	Confign. <sup>C)</sup>
l <sup>d)</sup>	la	A	43	50	s
2	la	в	54	69	S
3 <sup>d)</sup>	lb	В	49	36	S
4	lc	A	26	83 <sup>e)</sup>	S
5	lc	в	48	86 <sup>e)</sup>	S

 Table I
 Asymmetric Michael addition of diethyl malonate

a) A : KO-tert-Bu(1.2 eq)-CH<sub>2</sub>(COOEt)<sub>2</sub>(1.5 eq), B : KO-tert-Bu(1.5 eq)-CH<sub>2</sub>(COOEt)<sub>2</sub>(3.0 eq). b) Isolated yield. c) Based on  $[\alpha]_D^{20}$ +16.8°(benzene) for optically pure (S)-3. d) Partial racemization of the recovered amino acid ester was observed. e) Corrected for 93.5% optical purity<sup>1)</sup> of L-4c used.

In all cases examined, the optically active aldehyde 3 obtained by the present method using L-4 as a chiral reagent has the S configuration. This means that the stereochemical course of the present Michael reaction can be predicted by the attack of malonate anion from the less-hindered side of the S-trans conformer of the cyclic complex as shown in 2. It should be mentioned that in cases where tert-leucine tert-butyl ester 4c was used as a chiral reagent, the optical yields of 3 were found to be reasonably high (runs 4 and 5), and 4c was recovered without any loss of optical purity for reuse.

Studies on detailed mechanism of this reaction and its applications are in progress.

## References

- 1) S. Hashimoto, S. Yamada, and K. Koga, <u>J. Am. Chem. Soc.</u>, <u>98</u>, 7450 (1976).
- 2) For some examples of asymmetric Michael reaction, see a) G. Tsuchihashi,
- S. Mitamura, S. Inoue, and K. Ogura, <u>Tetrahedron Letters</u>, 323 (1973); b) H. Wynberg and R. Helder, <u>Ibid.</u>, 4057 (1975).
- E. D. Bergmann, D. Ginsberg, and R. Pappo, "Organic Reactions", Volume 10, J. Wiley & Sons, New York, <u>1959</u>, p. 179-555.
- 4) K. B. Wiberg and T. W. Hutton, <u>J. Am. Chem. Soc.</u>, <u>78</u>, 1640 (1956).